

A 2:1:1 2-(2-carboxyphenylamino)isophthalic acid–ethyl acetate–butan-2-one inclusion complex

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Key indicators

Single-crystal X-ray study
T = 295 K
Mean $\sigma(\text{C}-\text{C}) = 0.002 \text{ \AA}$
H-atom completeness 58%
Disorder in solvent or counterion
R factor = 0.052
wR factor = 0.167
Data-to-parameter ratio = 13.6

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

In the title compound, $\text{C}_{15}\text{H}_{11}\text{NO}_6 \cdot 0.5\text{C}_4\text{H}_8\text{O}_2 \cdot 0.5\text{C}_4\text{H}_8\text{O}$, extensive hydrogen bonding establishes a rigid host matrix in which a disordered mixture of ethyl acetate and butan-2-one solvent molecules is held.

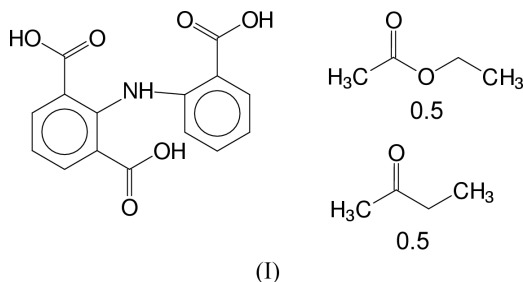
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Comment

The host compound in the title inclusion complex, (I), forms a robust supramolecular framework in the crystal structure *via* intramolecular and intermolecular hydrogen bonding (Fig. 1 and Table 1). A characteristic intramolecular bifurcated hydrogen bond establishes a V-shaped molecular skeleton, reducing the available hydrogen-bond functions towards the exterior. A hydrogen-bonding link to the keto O atom of the guest molecules establishes an important contact site for guest binding in the crystal structure (Fig. 2). A reduced melting point (519–521 K *versus* the reported 531–532 K for the pure compound; Rewcastle & Denny, 1985) may also be attributed to solvent encapsulation. As indicated by difference electron density maps, successful refinement and hydrogen-bonding geometry, all ionizable groups remain in the neutral form. As also demonstrated by a subsequent NMR analysis the crystals, originally recrystallized from butan-2-one, still retained about 20% butan-2-one in the solid after recrystallizing the first batch from ethyl acetate. A 1:1 ethyl acetate–butan-2-one proportion was adopted in the final structure model, omitting H atoms of the disordered guests. The guest molecules appear to be statistically distributed over a single crystallographically unique site, with the two components related by a local pseudo-twofold rotation axis. The packing involves an undulating host framework, with guests held in channels running parallel to the crystallographic *c* axis (Fig. 2).



Experimental

2-(2-Carboxyphenylamino)isophthalic acid was prepared as reported previously (Rewcastle & Denny, 1985), and was obtained from solution in butan-2-one. It was subsequently recrystallized from ethyl acetate by slow evaporation of the solvent, yielding large transparent crystals.

Crystal data

$C_{15}H_{11}NO_6 \cdot 0.5C_4H_8O_2 \cdot 0.5C_4H_8O$
 $M_r = 381.35$
 Monoclinic, $P2_1/c$
 $a = 7.487$ (1) Å
 $b = 16.444$ (1) Å
 $c = 15.748$ (1) Å
 $\beta = 100.21$ (1)°
 $V = 1908.1$ (3) Å³
 $Z = 4$

$D_x = 1.327$ Mg m⁻³
 Cu $K\alpha$ radiation
 Cell parameters from 25 reflections
 $\theta = 37.0$ – 39.8 °
 $\mu = 0.88$ mm⁻¹
 $T = 295$ (2) K
 Prism, colourless
 $0.56 \times 0.55 \times 0.43$ mm

Data collection

Enraf–Nonius CAD-4 diffractometer
 ω - 2θ scans
 Absorption correction: none
 4100 measured reflections
 3834 independent reflections
 3372 reflections with $I > 2\sigma(I)$
 $R_{int} = 0.012$

$\theta_{max} = 75.7$ °
 $h = -9 \rightarrow 9$
 $k = 0 \rightarrow 20$
 $l = 0 \rightarrow 19$
 3 standard reflections every 100 reflections
 intensity decay: 1%

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.052$
 $wR(F^2) = 0.167$
 $S = 1.10$
 3834 reflections
 281 parameters
 H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.1071P)^2 + 0.2617P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{max} < 0.001$
 $\Delta\rho_{max} = 0.44$ e Å⁻³
 $\Delta\rho_{min} = -0.22$ e Å⁻³
 Extinction correction: *SHELXL97*
 Extinction coefficient: 0.0052 (8)

Table 1

Hydrogen-bonding geometry (Å, °).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
$O2-H2O \cdots O7^i$	0.87	1.78	2.645 (2)	175
$O4-H4O \cdots O1^{ii}$	0.98	1.83	2.6576 (18)	140
$O5-H5O \cdots O6^{iii}$	0.83	1.83	2.6441 (19)	166
$N1-H1N \cdots O6$	0.98	1.96	2.7047 (16)	131
$N1-H1N \cdots O3$	0.98	2.23	2.7388 (19)	111

Symmetry codes: (i) $1-x, \frac{1}{2}+y, \frac{3}{2}-z$; (ii) $1-x, y-\frac{1}{2}, \frac{3}{2}-z$; (iii) $1-x, 1-y, 2-z$.

H atoms bonded to O and N were found in difference maps, while those bonded to C were positioned geometrically; they were all refined with a riding model, with $C-H = 0.93$ Å, $N-H = 0.98$ Å and $O-H = 0.83$ – 0.98 Å, and with $U_{iso}(H) = 1.3U_{eq}$ of the parent atom. H atoms were not located for the disordered solvent guest molecules, which were found to be disordered over a single site. A 1:1 distribution of the two guest molecules was assumed. No constraints or restraints were applied, and the geometry of the guest molecules is unreliable.

Data collection: Locally modified *CAD-4 Software* (Enraf–Nonius, 1989); cell refinement: *SET4* (de Boer & Duisenberg, 1984); data reduction: *XCAD4* (Harms, 1996); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *PLATON* (Spek, 2003); software used to prepare material for publication: *PLATON*.

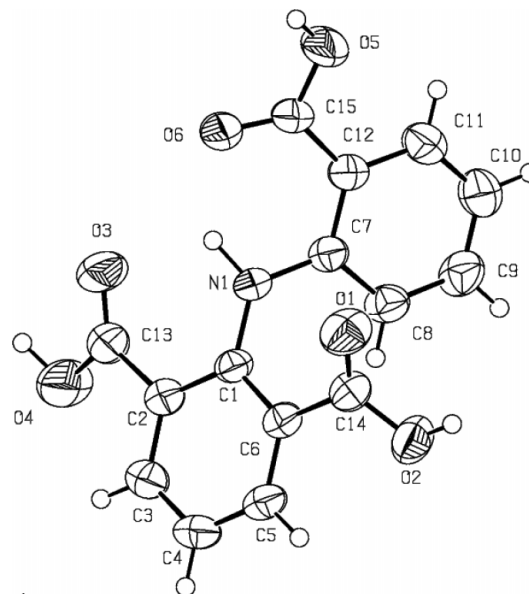


Figure 1

The molecular structure, with displacement ellipsoids drawn at the 50% probability level. The disordered solvent molecules are not shown.

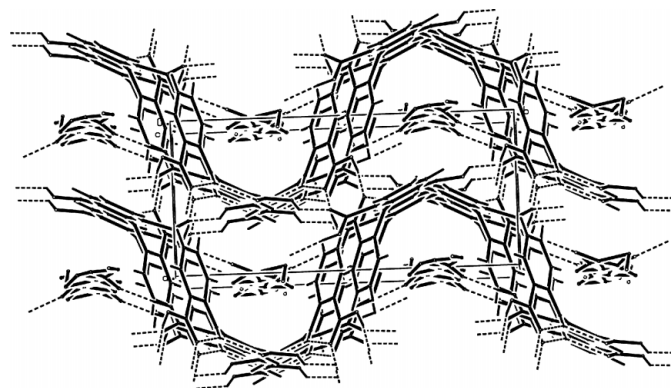


Figure 2

Packing diagram, showing the hydrogen-bonded host matrix with disordered guest molecules bound in channels parallel to the c axis. Hydrogen bonds are shown as dashed lines.

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References

- Boer, J. L. de & Duisenberg, A. J. M. (1984). *Acta Cryst.* **A40**, C-410.
 Enraf–Nonius (1989). *CAD-4 Software*. Version 5. Enraf–Nonius, Delft, The Netherlands.
 Harms, K. (1996). *XCAD4*. University of Marburg, Germany.
 Rewcastle, G. W. & Denny, W. A. (1985). *Synthesis*, pp. 217–222.
 Sheldrick, G. M. (1997). *SHELXS97* and *SHELXL97*. University of Göttingen, Germany.
 Spek, A. L. (2003). *J. Appl. Cryst.* **36**, 7–13.